

Regents Periodic Table Questions

Regents Periodic Table Questions - Periodic Table of the Elements C 12.011 -4 +2 +4 6 2-4 Atomic Mass Symbol Atomic Number Electron Configuration Selected Oxidation States Relative atomic masses are based on 12 C = 12.000 Note: Mass numbers in parentheses are mass numbers of the most stable or common isotope.

KEYChemistry : Periodic Table. Where available, study the preparation material through the Info Page, flash cards, hangman, multiple-choice and match the column formats. Then, test your recall of the material through the fill in the blanks format. In some activities, reverse quiz is available where questions become the answers, and answers become the questions.

22) Metals outnumber nonmetals on the periodic table because most of the elements a) are close to having a full octet b) have 1 to 2 valence electrons c) have a high ionization energy d) have a high electronegativity

23) As we move from left to right on the periodic table, the reactivity of metals a) increases b) decreases c) stays the same

Note Packet - Unit 3: Periodic Table (KEY) Here's the key to the notes for the unit. Use this to complete the homework assigned, and be sure to watch the vodcasts! ... Practice Regents Multiple Choice Questions - Periodic Table Use this link to access interactive multiple choice questions from regentsprep.org.